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16 terms Boyles law $P_1 \times v_1 = P_2 \times v_2$, Charles law $V_1/t_1 = V_2/t_2$, Gay lussacs law $P_1/t_1 = P_2/t_2$, Combined gas law $P_1 \times v_1 \text{ over } t_1 = P_2 \times v_2 \text{ over } t_2$

<http://quizlet.com/76584016/chemistry-gas-test-study-guide-flash-cards/>

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<http://www.ck12.org/chemistry/Solid-Liquid-and-Gas/studyguide/user:13IntC/States-of-Matter/>

Study Guide for AP Chemistry Chapter 5, Gas Laws . Students should be able to Explain the kinetic theory of matter as it applies to gases.

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